#### PATENT ABSTRACTS OF JAPAN

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#### (54) BLOOD SUGAR LEVEL DEPRESSING AGENT

#### (57) Abstract:

PURPOSE: To provide a blood sugar level depressing agent containing a compound such as 4-methoxy-N-3-pyridylbenzamide, etc. as an active component, and having excellent blood sugar level depressing effect and long duration of the activity.

CONSTITUTION: The agent contains the compound of formula[R<sub>1</sub> is H or lower alkyl; R<sub>2</sub> is straight-chain, branched-chain or cyclic alkyl, (nuclear-substituted) pyridyl, or pyridylmethyl; n is 1W3]as an active component. The active compound of formula can be prepared easily by reacting an amine with a methoxybenzoyl chloride in the presence of a base such as triethylamine by conventional process. It is administered in an arbitrary form prepared by the conventional means for the preparation of ordinary drug preparation.

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$$CON \begin{pmatrix} R_1 \\ R_2 \end{pmatrix}$$

# (19) 日本国特許庁 (JP)

①特許出願公開

# ⑩ 公開特許公報(A)

昭58—69812

©Int. Cl.<sup>3</sup> A 61 K 31/16 31/44 # C 07 D 213/40 識別記号 ADP 庁内整理番号 6408-4C ③公開 昭和58年(1983)4月26日発明の数 1審査請求 未請求

213/40 7138—4 C 213/75 7138—4 C

(全 5 頁)

# 60血糖降下剤

②特 願 昭56—167934

②出 願 昭56(1981)10月22日

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最終頁に続く

明細書

1. 発明の名称

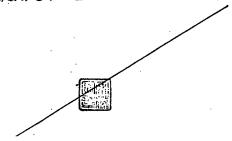
机糖降下剂

2. 特許請求の範囲

一般式

$$CON \begin{pmatrix} R_1 \\ R_2 \end{pmatrix}$$

(式中、R1 は水素原子又は低級アルキル基を示し、R2 は直鎖,分岐鎖又は環式アルキル基、核に置換券を有し得るピリジル基又はピリジルメテル基を示し、 e は 1 ~ 3 を示す。) で表わされる化合物を有効成分とする血糖降下剤。



3. 発明の詳細な説明

本発明は、次の一般式

$$(OCH_3)_n$$
 (1)

(式中、R1 は水素原子又は低級アルキル基を示し、R2 は直鎖,分岐鎖又は環式アルキル基、核に置換基を有し得るピリジル基又はピリジルメテル基を示し、nは1~3を示す。) で表わされる化合物を有効成分とする血糖降下剤の発明である。

上式 [1] で表わされる化合物の中には、公知の化合物が含まれるが、それらの配載されている先行文献には血糖降下作用ないしそれを示唆する楽 理作用は全く記載されていない。

上式 (1) で表わされる本発明の化合物は、例えば、以下の参考例に示すように、 アミン類とメトキシペンソイルクロライド類とを、 塩差、例えばトリエチルアミンの存在下常法により反応させることにより容易に得ることができる。

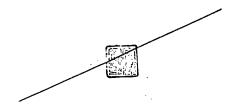
#### 参考例.

3 - アミノビリジン 9.4 9 ・トリエチルアミン 1 5 ml 及びアセトン 2 0 0 ml の混合解液に、氷冷機律下、4 - メトキシペンゾイルクロライド 1 7 9 を徐々に加える。同温度で 3 0 分、次いで家園で 1 時間機律後反応溶液を 1 M の水に注ぎ、折出する結晶を評取し、水洗後メタノールから再結晶して無色針状晶の 4 - メトキシー N - 3 - ビリジルペンズアミド(化合物 1) 1 7.5 9 を得た。収率 7 7 % ・融点 1 6 8 ~ 1 7 0 ℃

元素分析値 分子式 C12 H12 N2O2 として

C H N

理論値(%) 68.41 5.30 12.27 実測値(%) 68.33 5.27 12.24 上記と同様にして表1の化合物を得た。



					<b>継</b> 点	収塞	元	常分	析	領
Ab	-( OMe )=	Rı	R <sub>2</sub>	分子式	(2)	(%)	理論値(%) 実御値(%)	0	н	N
2	2-0Ne	н	$\bigcirc$	O <sub>13</sub> H <sub>12</sub> N <sub>2</sub> O <sub>2</sub>	112~114	7 6	6 8.4 1 6 8.4 9	5.3 0 5.2 4		1 2.2 7 1 2.3 1
3	•	•	Q <sub>Me</sub>	O14H14N2O3	80~82	8 3	6 9.4 0 6 9.3 2	5.83 5.80		1 1.5 6 1 1.5 <b>9</b>
. 4 .	,	•	Ç, m	O <sub>15</sub> H <sub>16</sub> N <sub>2</sub> O <sub>2</sub>	85~87	9 1	7 0.2 9 7 0.2 4	6.29 6.23		1 0.9 3 1 0.9 9
5	3-0Me	٠.		O13H12N2O2	121~122	8 5	6 8.4 1 6 8.4 8	5.3 0 5.3 6		1 2.2 7 1 2.2 1
. 6	•	•		•	155~156	8 3	6 8.4 1 6 8.4 3	5.3 0 5.3 1		1227
7	. ,	•	□ N <sub>Me</sub>	O14H14N2O2	99~101	8.8	6 9.4 0 6 9.4 7	5. 8 3 5. 7 9		1 1.5 6 1 1.6 0
8	4-0Me	•		O13H12N2O2	131~132	7 9	6 8 4 1 6 8 3 5	5.3 ( 5.2 (		1227
9	•	•	- CH2 (N)	O14H14N2O2	150~153	6 5	6 9.4 0 6 9.3 6			1 1.5 6 1 1.5 2
1 0	,	,	- cm2-	,	71~73	6 8	6 9.4 0 6 9.4 7			1 1.5 6 1 1.5 8
1 1	•	,	-CNL H€	•	61~64	7 7	6 9.4 0 6 9.4 5			1 1.5 6 1 1.6 3
1 2		,	Ön.	O15H15N2O2	136~137	8 2	7 0.2 9 7 0.3 7			1 0.9 3 1 0.8 9

				·	*.				
13	2,3-(OMe) <sub>2</sub>	н		O <sub>14H14</sub> N <sub>2</sub> O <sub>3</sub>	117~118	5 8	6 5.1 0 6 5.1 4	5.4 6 5.4 9	1 0.8 5 1 0.9 1
1.4	,	.,	In.	O15H16N2O2	110~111	6 2	6 6.1 6	5.9 2	1 0.2 9
			N. Me	018.11811203	110 111	0.2	6 6.1 2	5.95	, 1 0.3 3
15	,			0	l		6 7.1 1	6. 3 4	9.78
1.3			→ <sub>N</sub> ≠ He	. C1 6H1 8N 2O3	111~112	6 7	6 7.1 4	6.37	9.75
	0 4 60 K X	_					66.16	5.92	1 0. 2 9
16	2,4 - (OMe) <sub>2</sub>	•	- cH2 - N	O15H16N2O3	98~99	5 1	6 6.1 1	5.87	1 0.3 4
							6 6.1 6	5.92	1 0.29
1 7		,	- ne		140~141	6 9	6 6.2 1	5.96	1 0.3 1
18		_					6 7.1 1	6.34	9.78
1.6		.,	₩ Me	O16H18N2O3	93~94	6 3	6 7.1 5	6.39	9.74
							6 6.1 6	5.9 2	1 0.29
1 9	2,6 - (OMe)2	,	N Me	O15H16N2O3	155~156	67	66.22	5.97	1 0.24
	_	_	1 3.	_			- 6 7.1 1	6.34	9.78
2 0		,	L <sub>N</sub> L <sub>He</sub>	O16H16N2O3	206~209	63	67.07	6.39	9.80
		_		<u>.</u>			6 5.1 0	5.4 6	1 0.8 5
2 1	3,4 - (OMe)2	,		O14H14N2O3	84~86	7 9	6 5.1 6	5.41	1 0.87
• •							6 5.1 0	5.4 6	1 0.85
2 2		,		•	49~51	8 8	6 5.0 8	5.43	1 0.8 8
				0 " " 0			6 6.1 6	5.92	1 0.29
2 3	,	,	-cH2-	O16H16N2O3	1 2 2~1 2 3	6 3	6 6.1 2	5.97	1 0.24
24	_	,	-cm2-	_		7 4	66.16	5.9 2	1 0.2 9
	,		N. J.		128~129		6 6.1 9	5.88	1 0.3 3
2 5							6 6.1 6	5.9 2	1 0.29
			- MMe		131~132	7 5	6 6.20	5.96	1 0.2 5

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	7		r	<del></del>	· · · · · · · · · · · · · · · · · · ·				
2 6	3,4 - (O Me) <sub>2</sub>	н	He He	C16H18N2O3	69~71	63	6 7.1 1 6 7.1 5	6.3 4 6.3 7	9.7 8 9.7 7
2 7	,	,	i-Pr	O12H17NO3	144~145	8 5	6 4.5 5 6 4.5 9	7.68 7.61	6.27
2 8	,	,	n-Bu	O13H19NO3	83~84	88.	6 5.8 0 6 5.7 8	8.0 7 8.0 3	5.9 0 5.8 4
2 9	,	•	s-Bu	,	1 2 7~1 2 8	8 3	65.80	8.07	5.90 5.93
3 0	•	•	i -Bu	,	124~125	80	6 5.8 0 6 5.8 5	8. 0 7 8.1 1	5.9 0 5.9 5
3 1	•	,	-(H)	O <sub>15</sub> H <sub>21</sub> NO <sub>3</sub> -	181~182	9 1	6 8.4 1	8.04	5.3 2 5.3 6
3 2.	3,5-(OMe)2	,	□N Me	O <sub>15</sub> H <sub>16</sub> N <sub>2</sub> O <sub>3</sub>	96~97	8 5	66.16	5.9 2 5.9 8	10.29 10.32
3 3 .		,	O Me	C16H18N2O3	119~120	8 7	6 7.1 1 6 7.1 8	6.34	9.78
3 4	3,4,5-(OMe)3	,	0	O <sub>15</sub> H <sub>16</sub> N <sub>2</sub> O <sub>4</sub>	154~156	6 5	6249	5.5 9 5.6 4	9.7 2 9.7 1
3 5	,	,		,	157~158	7 7	6249	5.5 9	9.7 2 9.7 3
3 6		,	-042	O16H18N2O4	115~116	5 8	6 3.5 6 6 3.5 2	6.00	9.27
3 7	,	,	- cu <sub>2</sub> - Cu <sub>2</sub>	•	145~146	6 9	6 3.5 6 6 3.5 1	6.00	9. 2 5 9. 2 7 9. 2 2
3 8	,	,	√N ne	•	127~128	6 4	6 3 5 6 6 3 5 9	6.00	9.27 9.29

3 9	3,4,5-(OMe)	н	A ne	O <sub>17</sub> H <sub>20</sub> N <sub>2</sub> O <sub>4</sub>	145~146	7 1	6 4.5 4 6 4.5 8	6.3 7 6.3 2	8.8 6 8.9 0
4 0	, .	•	n-Pr	O13H19NO4	114~115	7 3	6 1.6 4 6 1.6 0	7.5 6 7.5 9	5.5 3 5.5 7
4 1	,	,	i-Pr	•	154~155	77	6 1.6 4 6 1.6 6	7.5 6 7.5 4	5.5 3 5.5 8
4 2	. ,	,	# - Bu	O14H21NO4	133~134	80	6 2 9 0 6 2 8 7	7.9 2 7.8 6	5. 2 4 5. 2 7
4 3	,	,	s-Bu	,	162~163	7 5	6 2 9 0 6 2 9 5	7.9 2 7.9 4	5. 2 4 5. 2 0
4.4	•	,	t - Bu	, .	133~134	7 9	6290 6291	7.9 2 7.8 8	5. 2 4 5. 2 9
4 5	•	•	i-Bu	,	1 2 2~1 2 3	8 1	6290 6296	7.9 <b>2</b> 7.8 7	5.24 5.28
4 6	•	,	-{H}	O16H23NO4	182~183	88	6 5.5 1 6 5.5 4	7.9 0 7.9 3	4.7 8 4.7 2
4 7	•	i-Pr	i-Pr	O16H25NO4	127~128	7 2	6 5.0 6 6 5.1 1	8.5 3 8.5 9	4.7 4 4.7 1

とのようにして得られる本発明の化合物は、優れた血・解降下作用を有し、ヒトに対しては 0.1~100 m/ 以で有効で、1日1回 0.1~100 m/ ノロの投与で 2.4 時間以上その効力を持続する。

投与に際しては、通常の製制化に用いられる慣用手段により所報の制型に成形された製剤が用い られる。

### 奥施例 1.

1 群 5 匹の 5 週令 D D Y 系マウス(雄,体質 2 5 ~ 3 0 9 )を 1 6 時間絶食後、 アロキサン 7 5 呼 / タを静脈内に投与し、 4 8 時間後に、 本発明化合物( 2 0 0 呼 / タ)の水溶液又はけん濁液を経口投与し、 1 5 0 分後に心臓から採血し、 グルコースオキンダーゼ法により 血中糖量を制定した。 倒定結果を表 2 に例示する。

なお、表中の化合物裕号は、移考例の化合物番号に対応している。

後 2

	血糖值(四/山)
投与化合物	mean ± S. D.
なし(対照)	47 #± 28
1	3 2 6 ± 4 2 **
3 .	3 7 8 ± 3 1 ••
4	3 6 4 ± 1 9 •••
6	3 7 8 ± 5 2 •
. 7	4 1 2 ± 3 3 •
1 2	3 8 3 ± 2 8 ••
1 7	3 4 5 ± 4 1 ***
2 2	378 ± 37 ••
2 5	3 5 5 ± 4 6 ••
2 6	3 3 6 ± 3 2 ***
2 7	407±30 +
2 8	402 ± 24 **
2 9	4 2 1 ± 2 7 • ;
3 2	4 1 6 ± 2 3 •
-3 3	402 ± 34 •
3 6	4 1 6 ± 2 1 ••
3 8	3 0 7 ± 4 3 •••
3 9	4 1 2 ± 3 1 •
4.1	4 2 1 ± 2 8 •
4 6	3 8 3 ± 4 1 ••

\* : P < 0.05 . \* \* : P < 0.01 . \* \* \* : P < 0.001

#### 夹筋侧 2

4-メトキシ-N-3-ビリジェル

ペンメアミド(化合物1)



リン酸水岩カルシウム

8.5部

結晶セルロース

5 0 <del>8</del>8

コーンスターチ

4 0 4

ステアリ酸カルシウム

1. 5 部

とれらをよく混合し、常法により1錠250mm に打錠(有効成分100mm含有)し、血糖降下用 錠剤として用いる。

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# 第1頁の続き

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#### **DRAFT TRANSLATION**

from

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### **JAPANESE PATENT APPLICATION (A)**

No. 58-069812

### A HYPOGLYCEMIC AGENT

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Please Note- Names of Japanese firms, research laboratories and government entities, as translated are not necessarily identical with the names adopted by such organisations for international contacts. Japanese personal and surnames often permit of several readings and the ones used in this translation are not necessarily the ones preferred by their bearers. Foreign names mentioned in Japanese specifications cannot always be accurately reconstructed.

J58-069812
(unexamined)

# Caution: Translation Standard is Draft Translation

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Examination request: Not yet made

Number of Inventions: 1

(Total 5 pages)

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31/44		
// C07D 213/40		7138-4C
213/75		7138-4C

#### **Specification**

### 1. Title of Invention

Hypoglycaemic agent

#### 2. Patent Claim

Hypoglycemic agent which has a compound represented by the following formula as the active component.

[In the formula,  $R_1$  denotes hydrogen atom or lower alkyl group,  $R_2$  denotes a linear, branched or cyclic alkyl group, a pyridyl group which may have a substituent on the nucleus or a pyridylmethyl group, and n denotes 1-3].

## 3. Detailed Description of the Invention

This invention is the invention of a hypoglycemic agent which has a compound represented by the following formula (I) as the active component

$$CON < R_2$$
 $(OCH_3)_n$ 
[I]

[In the formula,  $R_1$  denotes hydrogen atom or lower alkyl group,  $R_2$  denotes a linear, branched or cyclic alkyl group, a pyridyl group which may have a substituent on the nucleus or a pyridylmethyl group, and n denotes 1-3].

Known compounds are included in the aforesaid compound represented by the formula (I), but in the previous literature in which they are mentioned, there is no mention at all of a hypoglycemic effect or a pharmacological action suggesting this.

The compounds of this invention represented by the aforesaid formula (I) may be obtained readily by usual methods of reacting an amine compound with a methoxybenzoyl chloride compounds in the presence of a base such as triethylamine, as illustrated in the following reference example.

# Reference Example

4-methoxybenzoyl chloride 17 g was added gradually under ice cooling and stirring to a mixed solution of 3-aminopyridine 9.4 g, triethylamine 15ml and acetone 200 ml. After stirring for 30 minutes at the same temperature then for 60 minutes at room temperature, the reaction solution was poured into 1 l of water, and the crystals which precipitated were collected by filtration and washed with water, then re-crystallised from methanol, to obtain 175 g of colourless acicular crystals of 4-methoxy-N-3-pyridylbenzamide (compound 1), melting point 168-170°C.

Elemental analysis	as molecular formula $C_{13}H_{12}N_2O_2$						
	C	H	N				
theoretical value (%)	68.41	5.30	12.27				
experimental value (%)	68.33	5.27	12.24				

The compounds of Table 1 were obtained in the same way.

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CON (R <sub>2</sub>	
(0 654) <sub>3</sub>	

Table 1

			Ro	Molecular	Melting	Yie	ld Eleme	ental a	nal. valı	ıes
No.	-(OMe)n	$R_1$	$R_2$	formula	point		Calc	(%) (	CH	N
					(°C)	(%	6) Found	d(%) C	H	N
	2 011	н		0 N . O .	112~114	7 6	6 8.4 1	5.30	1 2 2 7	
2	2-ONe	"		O13H13N2O2			6 8.4 9	5, 2 4	1 2.3 1	╛
	,		Q <sub>n</sub>	014H14N2O2	80~82	8 3	6 9.4 0	5.83	1 1.5 6	1
3				014111411202			6 9.3 2	5.80	1 1.5 9	_
4	,	,	Ç, n.	O15H16N2O2	85~87	9 1	7 0.2 9	6.29	1 0.9 3	
			→ <sub>N</sub> .P.Hz				7 0. 2 4	6.23	1 0.9 9	4
5	3-0Ne	•		O13H12N2O2	121~122	8 5	6 8.4 1	5.30	1227	
-			~, <sub>N</sub> ,y				6 8.4 8	5.3 6	1 2 2 1	-
6	,		$\Diamond$	,	155~156	8 3	68.41	5.3 0 5.3 1	1227 1230	
			<b>\</b> <sub>N</sub> <i>y</i>							-
7		,	Q <sub>n</sub>	014H14N2O2	99~101	8 8	6 9.4 0	5. 8 <b>3</b> 5. 7 9	1 1.5 6 1 1.6 0	
<u> </u>		<b> </b>	NAM.				6841	5.30	1227	$\dashv$
8	4-OMe		-	O1 1H 1 2N 2O2	131~132	7 9	6835	5.26	1231	-
		<del> </del>	N.J		<b> </b>		6 9.4 0	5.8 3	1 1.5 6	$\dashv$
9	•	•	- CH,	014H14N2O2	150~153	6 5	6 9.3 6	5.79	1 1.5 2	
-		<del> </del>			<del> </del>		6 9.4 0	5.8 3	1 1.5 6	$\dashv$
10	•	•	- CM3-C)	•	71~73	68	6 9.4 7	5.78	1 1.5 8	1
}		<del>                                     </del>	_		<del> </del>	7 7	6 9.4 0	5.8 3	1 1.5 6	$\dashv$
11	•	•	Q.m.	•	61~64		6 9.4 5	5.8.8	1 1.6 3	Ì
1		1	1				7 0.2 9	6.29	1093	$\neg$
1 2	j ,	•	- In	O15H16N2O2	136~137	8 2	7 0.3 7	6.34	1 0.8 9	

1 3	2,3-(OMe) <sub>2</sub>	н		O14H14N2O3	117~118	5 8	6 5.1 0 6 5.1 4	5.4 6 5.4 9	1 0.8 5
14	•	,	In me	O16H16N2O3	110~111	6 2	66.16	5.9 2 5.9 5	1 0.2 9
1 5		•	<b>Ö.</b>	. O1 6H1 8N2O3	111~112	6 7	67.11	6.34	9.7 8 9.7 5
16	2,4 - (OMe)2	•	-04-	O15H16N2O3	98~99	5 1	66.16	5.9 2 5.8 7	1 0. 2 9
1 7		,	Q.ne	,	140~141	6 9	6 6.1 6 6 6.2 1	5.9 2 5.9 6	1 0.2 9
1 8	,	•	<u>ڳ</u>	O16H18N2O3	93~94	6 3	6 7.1 1 6 7.1 5	6.34	9.78
19	2,6 - (OMe) <sub>2</sub>	•	Q me	O15H16N2O3	155~156	6 7	6 6.1 6	5.92	10.29
2 0	•	•	Ö.	O16H18N2O3	206~209	63	67.11	6.34	9.78
2 1	3,4 -(ONe)2	•	<b>Q</b>	O14H14N2O3	84~86	7 9	6 5.1 0	5.4 6 5.4 1	1 0.8 5
2 2	,	,		,	49~51	8 8	6 5.1 0 6 5.0 8	5.4 6 5.4 3	1 0.8 5
2 3	,	,	-cu <sub>2</sub> -	O16H16N2O3	1 2 2~1 2 3	6 3	6 6.1 6 6 6.1 2	5.9 2 5.9 7	1 0.8 8
2 4	,	,	- cm2-CN	,	128~129	7 4	66.16	5.9 2	1 0.2 9
2 5	•	•	A. He	•	131~132	7 5	6 6.1 6 6 6.2 0	5.88 5.92 5.96	1 0.3 3

			<del></del>				
2 6	3,4 - (O Me) <sub>2</sub>	н	A He	O <sub>16</sub> H <sub>18</sub> N <sub>2</sub> O <sub>3</sub>	69~71	63	6 7.1 1 6.3 4 9.7 8 6 7.1 5 6.3 7 9.7 7
2 7	•	,	i-Pr	O13H17NO3	144~145	8 5	6 4.5 5 7.6 8 6.2 7 6 4.5 9 7.6 1 6.2 3
2 8	,	,	e-Bu	O13H16NO3	83~84	8 8	65.80 8.07 5.90 65.78 8.03 5.84
2 9	,	•	s-Bu	,	1 2 7~1 2 8	8 3	65.80 8.07 5.90 65.84 8.04 5.93
3 0	,	•	i-Bu	,	124~125	8 0	6 5.8 0 8.07 5.9 0 6 5.8 5 8.1 1 5.9 5
3 1	•	•	(F)	O <sub>15</sub> H <sub>21</sub> NO <sub>2</sub> .	181~182	9 1	6841 8.04 5.32 6836 8.07 5.36
3 2.	3,5 - (OMe);	•	- ne	O18H16N2O3	96~97	8 5	6 6.1 6 5.9 2 10.2 9 6 6.1 2 5.9 8 10.3 2
33.	,	,	M. Me	O16H18N2O3	119~120	8 7	6 7.1 1 6.3 4 9.7 8 6 7.1 8 6.3 7 9.7 2
3 4	3,4,5-(OMe)3	,		O18H14N2O4	154~156	6 5	6249 5.59 9.72 6253 5.64 9.71
3 5	•	,	Q	•	157~158	7 7	6 2 4 9 5.5 9 9.7 2 6 2 5 2 5.5 6 9.7 3
3 6		•	- OIL-	C16H18N2O4	115~116	5 8	6 3.5 6 6.0 0 9.2 7 6 3.5 2 6.0 4 9.2 5
3 7	•		-u, ()	,	145~146	6 9	6 3.5 6 6.00 9.2 7 6 3.5 1 6.0 7 9.2 2
3 8	,		√n <sub>e</sub>	,	127~128	6 4	6 3:5 6 6.0 0 9.2 7 6 3.5 9 6.0 3 9.2 9

3 9	3,4,5-(OMe)3	н	<u> </u>	O <sub>17</sub> H <sub>20</sub> N <sub>2</sub> O <sub>4</sub>	145~146	71	6 4.5 4 6 4.5 8	6.3 7 6.3 2	8.8 6 8.9 0
4 0	•	•	n-Pe	013H19NO4	114~115	73	6 1.6 4 6 1.6 0	7.5 6 7.5 9	5.5 3 5.5 7
4 1	•		i-Pr	,	154~155	77	6 1.6 4 6 1.6 6	7.5 6 7.5 4	5.5 3 5.5 8
4 2	,	•	#-Bu	O14H21NO4	133~134	80	6 2 9 0 6 2 8 7	7.9 2 7.8 6	5.24 5.27
4 3		•	e-Bu		162~163	75	6290 6295	7.9 2 7.9 4	5. 2 4 5. 2 0
4.4	•		t-Bu	•	133~134	7 9	6 2 9 0 6 2 9 1	7.9 2 7.8 8	5. 2 4 5. 2 9
4 5	•		i-Bu	,	122~123	8 1	6290 6296	7.9 <b>2</b> 7.8 7	5.24 5.28
4 6	•		-(H)	O16H18NO4	182~183	88	6 5.5 1 6 5.5 4	7.9 0 7.9 3	4.78
4 7	•	i-Pr	i-Pr	O16H28NO4	127~128	7 2	6 5.0 6 6 5.1 1	8.5 3 8.5 9	4.74

The compounds of this invention obtained in this way have excellent hypoglycemic action, and are effective at 100 mg/kg in man, and their effect is maintained by administration of 0.1-100 mg once a day for 24 hours or more.

For administration, a preparation is used which has been formed into the desired form by a customary means normally used in drug formulation.

# Lyample 1

5 week-old mice (male, body weight 25-30g) with 5 animals in a group were fasted for 16 hours, and then alloxan at 75 mg/kg was administered intravenously. After 48 hours, a solution or suspension of a compound of this invention (200 mg/kg) was administered of ally, and after 150 minutes, blood was taken from the heart and the glucose level was in casured using glucose oxidase. The measurement results are exemplified in Table 2.

Table 2

Administered	Blood glucose value (mg/dl)
compound	mean $\pm$ S.D.
None (control)	$473 \pm 28$
1	3 2 6 ± 4 2 ••
3	378±31 ••
4	3 6 4 ± 1 9 •••
6	378 ± 52 •
7	4 1 2 ± 3 3 •
1 2	3 8 3 ± 2 8 ••
1 7	3 4 5 ± 4 1 •••
2 2	3 7 8 ± 3 7 ••
2 5	355±46 ••
2 6	3 3 6 ± 3 2 •••
2 7	407±30•
2 8	4 0 2 ± 2 4 ••
2 9	4 2 1 ± 2 7 • ;
3 2	4 1 6 ± 2 3 •
3 3	402±34•
3 6	4 1 6 ± 2 1 ••
3 8	3 0 7 ± 4 3 •••
3 9	4 1 2 ± 3 1 •
4 1	4 2 1 ± 2 8 •
4 6	383±41 ••
F:P<0.05.**:P	<001 *** P<0001

 $\Gamma_{i}$  the Table, the compound number corresponds to the compound number of the reference examples.

# 1 xample 2

4-methoxy-N-3-pyridylbenzamide (compound 1)	100 parts		
calcium hydrogen phosphate	58.5 parts		
crystalline cellulose	50 parts		
corn starch	40 parts		
calcium stearate	1.5 narts		

These components were mixed well and pressed into 250 mg tablets (content of active component 100 mg/tablet) by usual methods, for use as a hypoglycemic agent.

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